

DOES THIS LOOK LIKE A WAY OF SAVING ENERGY?

**GENERATING LESS CARBON DIOXIDE FOR
TRANSPORTATION?**



**THE INTERNAL RESISTANCES IN COMMON BATTERIES
MOST OF THE “INTERNALS” OF A BATTERY ARE METALS
OR ELECTROACTIVE SEMI-CONDUCTORS.
BELOW ARE LISTED THE CONDUCTIVITIES OF THE USUAL
METALS AND SEMI-CONDUCTORS FOUND IN MODERN
BATTERIES AND ALSO THE USUAL ELECTROLYTES**

Metals (Resistance)

- Pb (lead) $22 \times 10^{-6} \text{ ohm/cm}^3$
- Ni (nickel) $7 \times 10^{-6} \text{ ohm/cm}^3$
- Fe (iron) $10 \times 10^{-6} \text{ ohm/cm}^3$
- Cd (cadmium) $7.6 \times 10^{-6} \text{ ohm/cm}^3$
- Cu (copper) $1.72 \times 10^{-6} \text{ ohm/cm}^3$

Semi-Conductors

- Generally $\sim 10^{-3} \text{ ohm/cm}^3$

Electrolytes

- H_2SO_4 (sulfuric acid) (5 M) $\sim 1 \text{ ohm/cm}^3$
- KOH (potassium hydroxide) $\sim 5\text{M}$
 $\sim 2 \text{ ohm/cm}^2$
- The internal resistance therefore exists
99% in the electrolyte.

**A USEFUL STRATEGY THEN,
WOULD BE TO TRY TO INCREASE
CONVECTION**

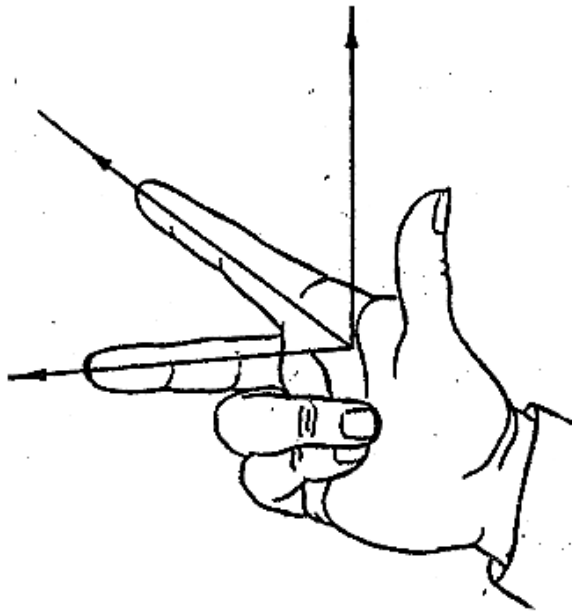
**MECHANICAL STIRRING HAS BEEN
TRIED – IN THE ROTATING DISK,
IN THE DROPPING MERCURY
ELECTRODE (ANALYTICAL
ONLY), AND BY PUMPING AS IN
ELECTROMACHINING**

RIGHT HAND RULE

Thumb (Current)

Forefinger(Field)

Centre finger (Motion)



The magnetic field (H) interacting with the current (upward flow of ions at the cathode (C) and downward at the anode (A)) gives a force into the cell and out of the cell causing precession of the convective path, lower viscosity and more rapid stirring.

**A USEFUL STRATEGY THEN,
WOULD BE TO TRY TO INCREASE
CONVECTION**

**THE DIRECTION OF
ROTATION OF AN ELECTRIC
MOTOR IS GIVEN BY THE
RIGHT HAND RULE, THAT IS
THE DIRECTION OF THE
MAGNETIC FIELD, THE
ELECTRIC CURRENT AND
THE DIRECTION OF
ROTATIONAL FORCE ARE
ALL MUTUALLY AT 90
DEGREES TO EACH OTHER.**

**THE MOTION OF THE ELECTROLYTE MUST
THEN BE ACROSS THE FACE OF THE
ELECTRODES**

**AT THE CATHODE THE NATURAL CONVECTION
IS NORMALLY UPWARD AS IONS ARE
DISCHARGED**

**ADDITION OF THESE TWO VECTORS RESULTS,
USUALLY, IN AN UPWARD SPIRAL AT THE
CATHODE AND A DOWNWARD ONE AT THE
ANODE**

**USING THE FRESNEL FORCE, AS IN THE
ROTATION OF AN ELECTRIC MOTOR, IS
MORE EFFICIENT IF THE ELECTRODES ARE
MAGNETIZED TO HAVE OPPOSITE
MAGNETIC POLES ON THEIR TWO FACES**

**WE BELIEVE THAT THE HIGH DRAIN WHEN
UNMAGNETIZED CAUSED SOME
DETERIORATION**

**IN THE ELECTROCHEMICAL
CELL, THE CONVECTING
STREAM OF IONS, EITHER
UP OR DOWN IS THE
ELECTRIC CURRENT – NOT
THE PASSAGE OF
ELECTRICITY FROM THE
FACE OF ONE ELECTRODE
TO THE OTHER**

**TO BE MUTUALLY
PERPENDICULAR THE
MAGNETIC FIELD MUST BE
FROM THE FACE OF ONE
ELECTRODE TO THE FACE
OF THE OTHER.**

MAGNETOHYDRODYNAMICS AS A GENERAL ELECTROCHEMICAL ENHANCER

**OF THESE THREE
PROCESSES, CONVECTION
AT VERTICAL ELECTRODES
WITH LIQUID ELECTROLYTE,
WILL DOMINATE IF THE
PROCESS IS DONE IN THE
EARTH'S GRAVITATIONAL
FIELD.**

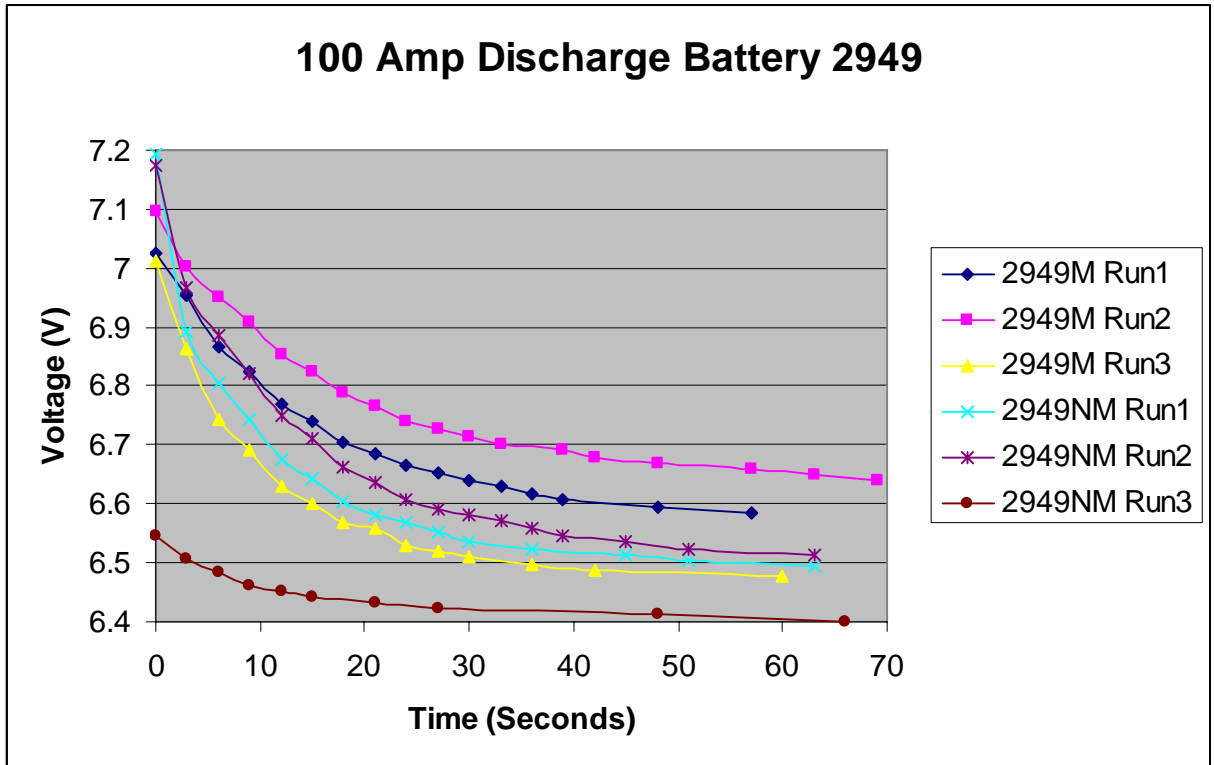
**IN SEVERAL EXPERIMENTS WITH
FLOATED PARTICLES,
ENHANCEMENT OF THE RATE
OF CONVECTION, OR
TRANSPORT OF ACTIVE IONS,
HAS BEEN OBSERVED TO BE
OVER TWO ORDERS OF
MAGNITUDE**

**THE RESULT WAS A VERY LARGE
DECREASE IN THE
ELECTROLYTE RESISTANCE,
OR THE LOSS OF ENERGY IN
OVERCOMING AN UN-
NECESSARY RESISTANCE IS
LARGELY ELIMINATED**

MAGNETOHYDRODYNAMICS AS A GENERAL ELECTROCHEMICAL ENHANCER

**IN ANY ELECTROCHEMICAL
PROCESS THE GENERAL DESIGN
PROCEDURE IS FIRST TO SELECT
ELECTRODES THAT ARE GOOD
CATALYSTS FOR THE REACTION
CONTEMPLATED.**

**THIS WILL MEAN THAT THE
ELECTROCHEMICAL PROCESS
RATE WILL NOT BE CONTROLLED
BY THE CHEMICAL REACTION
RATE, BUT BY THE TRANSPORT
PROCESSES – MIGRATION,
DIFFUSION, AND CONVECTION**



HIGH RATE DISCHARGE OF STANDARD PRODUCTION BATTERIES (15C)

MODULES FOR TRACTION BATTERIES OF A POPULAR BRAND NAME

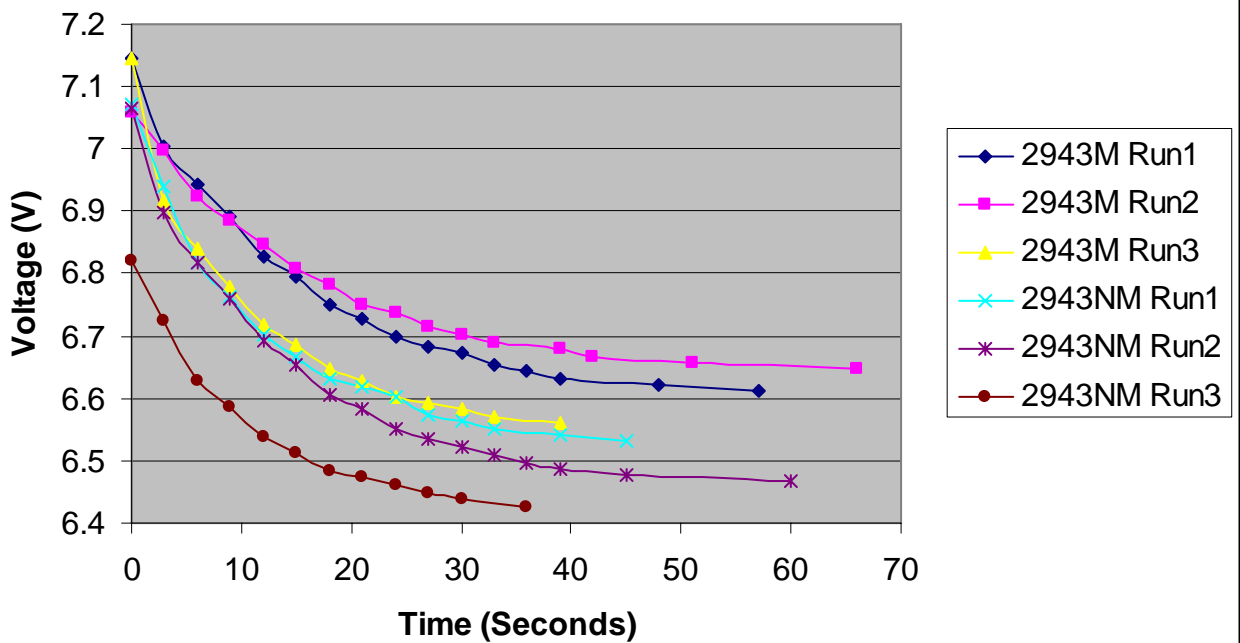
SUPPLIED BY A LARGE BATTERY MANUFACTURER

3 NiMH PRODUCTION MODULES TESTED BEFORE MAGNETIZATION AND AFTER

TESTED BY VIZON SCITEC VANCOUVER

2943

100 Amp Discharge Battery 2943



100 Amp Discharge Battery 2947

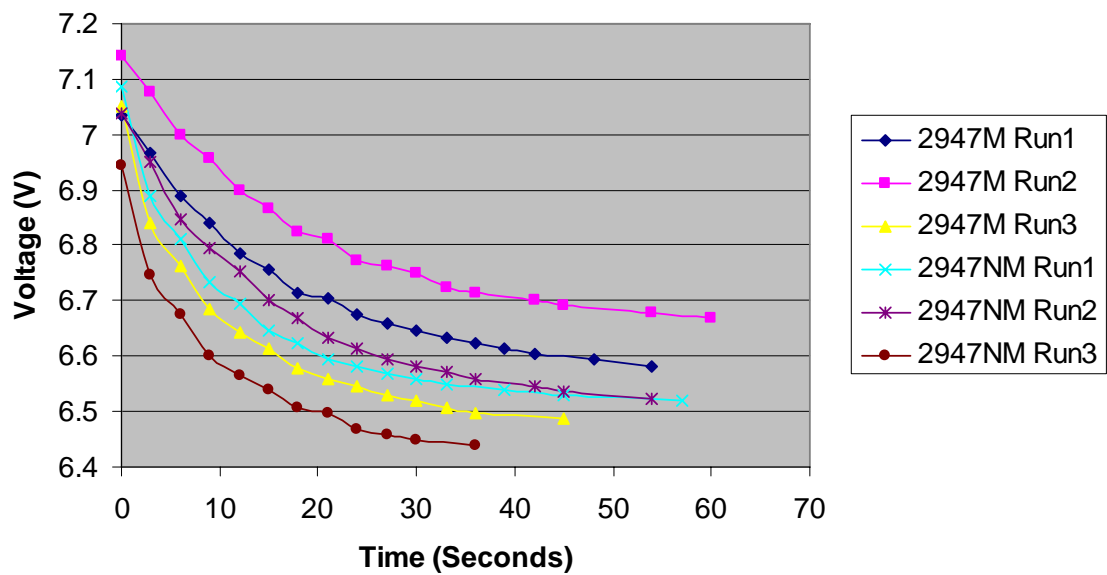


TABLE 1

COMPARISON OF PERFORMANCE OF NON-MAGNETIZED AND MAGNETIZED NiMH BATTERIES (3, 100 AMP RUNS)

BATTERY	RUN #	INCREASE	BATTERY	RUN#	INCREASE
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2943	1	16%	2947	1	13%
	2	28		2	23
	3	32		3	15

BATTERY	RUN #	INCREASE
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2949	1	15%
	2	18
	3	32

ARITHMETICAL AVERAGE OF THE THREE = 26.3%

MORE LIKELY AVERAGE IS ~ 32%

COMPARISON OF IMPROVEMENT ASSUMPTION/LIMITATIONS

- 1. WHEN A TEST WAS STOPPED AND RETESTED BEFORE ONE MINUTE, FOR THE NON-MAGNETIZED RUN, THE CURVE FOR A MAGNETIZED RUN WAS CONSIDERED TO HAVE STOPPED AT THE SAME TIME FOR FAIRNESS**
- 2. THIS PROCEDURE FOR FINDING THE AREA UNDER THE CURVE (WATT SECONDS) WAS POINTED OUT IN COMMENTS ON THE PLOTS FOR #2947, AND WAS USED FOR ALL PERCENTAGE INCREASES IN PERFORMANCE.**

COMPARISON OF IMPROVEMENT ASSUMPTION/LIMITATIONS

3. ONLY #2947 DID NOT INCREASE IN PERFORMANCE WITH EACH RUN, BUT ACTUALLY DECREASED USING THE “FAIRNESS” PROCEDURE.
4. NO ACCOUNT WAS TAKEN OF THE VOLTAGE TIME CURVE EXTENSION FOR THE MAGNETIZED RUNS IN EXCESS OF THE NON-MAGNETIZED RUNS – FURTHER INCREASE?
 $32 + ? = 40\% ??$

**TEN OF THESE MODULES MAKE UP
ONE TRACTION BATTERY FOR A
HYBRID AUTOMOBILE.**

**A 100 AMP DISCHARGE IS THEN 1000 AMP
CAPABILITY FOR ONE MINUTE ...
... A SEVERE TEST**

**THE REASON FOR STOPPING WAS
PRESUMED TO BE OVER-HEATING
AND/OR PRESSURE**

**NOTE THAT IT OCCURRED MUCH SOONER
FOR THE UN-MAGNETIZED RUNS WHICH
WERE DONE FIRST – THEN THE SAME
CELLS MAGNETIZED AND RETESTED**

**IN CHARGING A MAGNETIZED
BATTERY, ALMOST ALL OF THE
ENERGY APPLIED IS STORED IN THE
USUAL ELECTROCHEMICAL WAY,
AND ON DISCHARGE AGAIN, MOST
OF IT IS AVAILABLE FOR WORK**

**THE PLOTS OF THE DISCHARGE OF
THE NiMH BATTERIES SHOWED THIS
INCREASE IN PERFORMANCE. THE
NEXT PANEL SHOWS MHD APPLIED
TO THE MANUFACTURE OF NaClO_3
ELECTROCHEMICALLY**

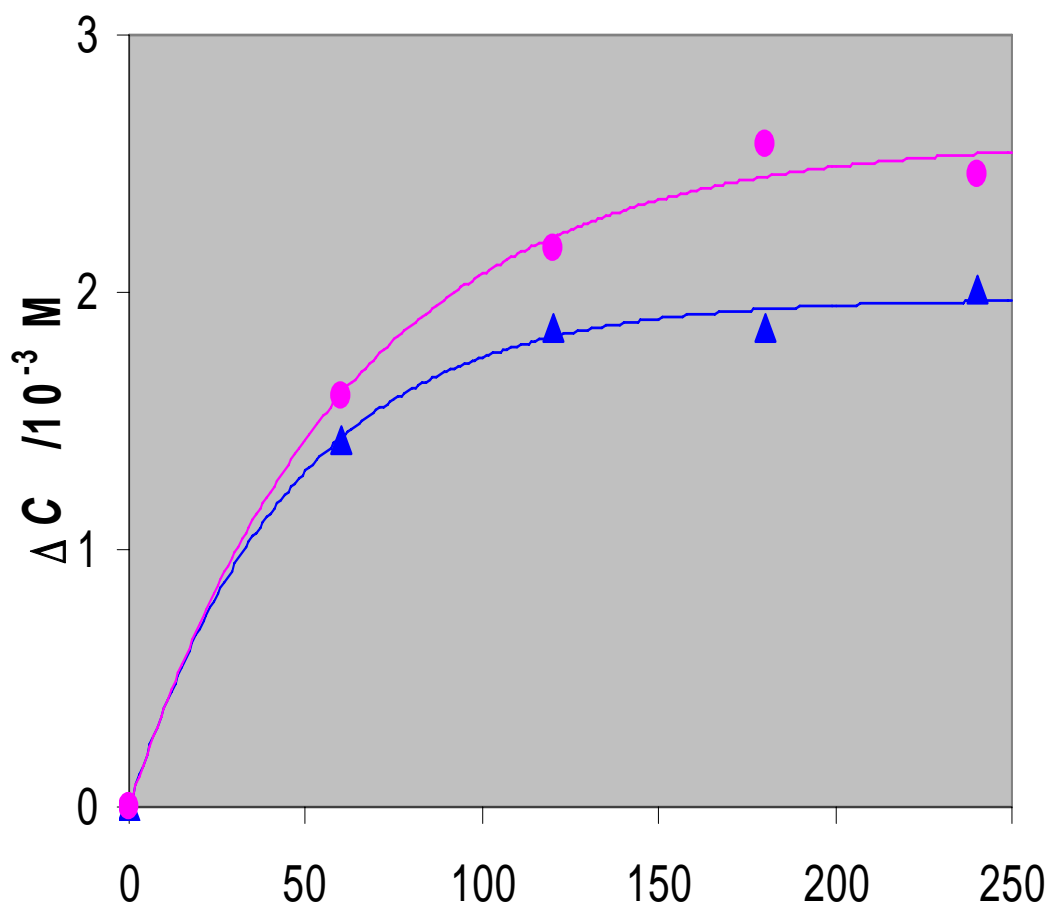
**FLOODED, TWO CELL ELECTRODE
BATTERIES MADE FROM SPARE
VARTA BATTERIES, ONE
PREVIOUSLY MAGNETIZED, ONE
NOT, WERE TESTED AND THE DRB
CAN. SCIENTISTS AND THE NRC CAN.
SCIENTISTS SAW RESULTS WHERE
THE MAGNETIZED BATTERY
PERFORMED ALMOST AN ORDER OF
MAGNITUDE BETTER, ALL AGAIN AT
FLEXIBLE SOLUTIONS' LAB**

**NOTE #2947 – THE TRIAL WAS ABANDONED AT
~ 35 SECONDS WHEN UNMAGNETIZED, BUT
WHEN LATER MAGNETIZED AND RETESTED
WENT 10 SECONDS LONGER AT THE HIGHER
VOLGAGE ... ~ 6.5 TO 6.45 V**

**CONSIDERING AGAIN 2947 (THE WORST
PERFORMER), AND COMPARING THE UN-
MAGNETIZED RUN WITH THE FIRST
MAGNETIZED RUN, THE FOURTH 100 AMP
DRAIN, THE MAGNETIZED RUN IS FLAT AT
ONE MINUTE AND 6.7 V - A REMARKABLE
RECOVERY.**

**THE OTHER TWO MODULES ARE ALSO
REMARKABLE IN THAT EACH MAGNETIZED
RUN IS BETTER THAN ANY UN-MAGNETIZED
RUN.**

THE INCREASED CONVECTIVE EFFECT OF INDIFFERENT PARAMAGNETIC IONS



NEW DIRECTION IN ELECTROCHEMISTRY

Energy saving magnetohydrodynamics

IN ALL ELECTROCHEMICAL PROCESSES, BATTERIES (INCLUDING NiMH, Li-ion) ELECTRODEPOSITION AND ELECTROPRODUCTION, THE DIRECTION OF RESEARCH AND APPLICATIONS IS IN THE DIRECTION OF LESS MOBILE ELECTROLYTE (BATTERIES) OR MORE ELECTROLYTE IN MOST OTHER PROCESSES (ELECTROWINNING, ETC.)

IN BATTERIES, THE TARGET OF “SHIPPING DRY” HAS LEAD TO LESS ELECTROLYTE, AND IN SOME CASES, ABSORPTION OF THE ELECTROLYTE IN A POROUS SEPARATOR. THIS INEVITABLY MEANS MORE RESISTANCE (WASTE OF ENERGY) ON CHARGING AND (WASTE OF ENERGY) ON DISCHARGING. ... For hybrids lower gas mileage)

Li ION BATTERIES

CHARACTERISTICS

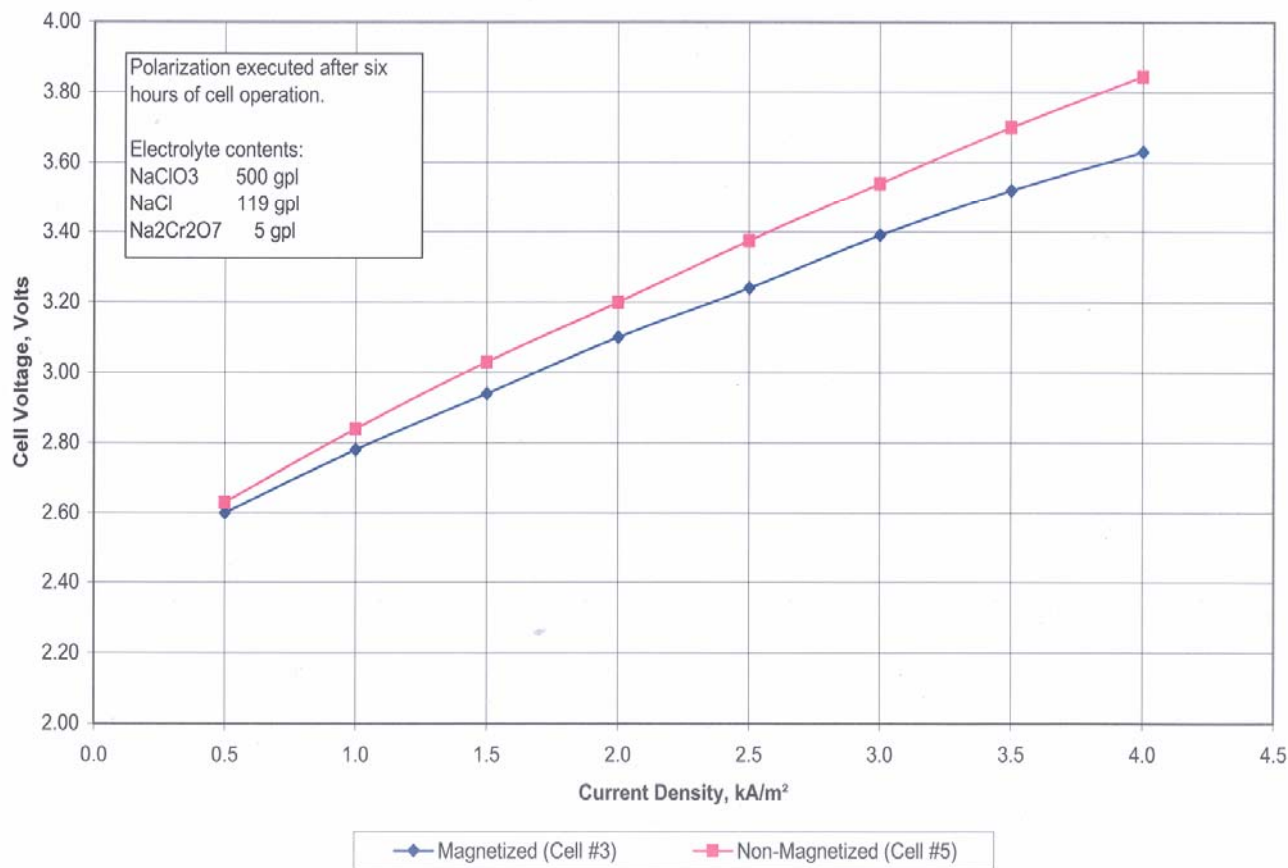
- **HIGH INTERNAL RESISTANCE OF ORGANIC ELECTROLYTE SOLUTIONS**
- **LEADS TO WASTAGE OF SUPPLIED ENERGY TO HEATING SOLUTION**
- **LEADS TO SLOW RECHARGING**
- **IN USE, WASTAGE OF STORED ENERGY IN HEATING THE SOLUTION IN AND OUT**
- **LEADS TO FIRE, EXPLOSION**

REMEDY

- **USE FERROMAGNETIC CORE FOR ELECTRODES (BOTH) AND MAGNETIZE (Fe, Ni OR Co)**
- **PLATE WITH REQUIRED METAL – (Cu, Al etc.) COAT WITH ELECTROACTIVES, ADD A PARAMAGNETIC INDIFFERENT ION OR A PERMANENT FREE RADICAL**
- **EXPECT REDUCTION IN INTERNAL RESISTANCE OF AT LEAST 30%**

Fig.1: POLARIZATION CURVE

Figure 1: Polarization Curve - Cell Voltage versus Current Density at 80°C
under natural circulation of electrolyte
(Table done by Aker Kvaener Chemetics, Vancouver BC)



THE LIMITATION ON CURRENT DENSITY AND HENCE PRODUCTION PER DAY IS THE AMOUNT OF OXYGEN PRODUCED.

THIS IS BECAUSE THE CATHODE IS PRODUCING ALMOST 100% HYDROGEN AND AT A CERTAIN CURRENT DENSITY OXYGEN PRODUCTION OCCURS AT THE ANODE.

THE EXPLOSION RANGE FOR PURE HYDROGEN IN PURE OXYGEN IS FROM ABOUT 4% TO ABOUT 94%, THE WIDEST RANGE OF ANY GAS. SO THE LIMIT OF THE CURRENT DENSITY IS WHEN THE UNWANTED OXYGEN BEING PRODUCED APPROACHES A PREDETERMINED RANGE LESS THAN 4% WHEN NO FURTHER VOLTAGE IS APPLIED.

**OTHER PROCESSES WHERE SOME
RESEARCH HAS BEEN DONE:
GENERATION OF HYDROGEN BY THE
ELECTROLYSIS OF WATER**

**IN EXPERIMENTS CONDUCTED IN
FLEXIBLE SOLUTIONS INTERNATIONAL
LABORATORY BY FLEXIBLE'S
TECHNICIANS AND WITNESSED BY
NATIONAL RESEARCH COUNCIL OF
CANADA'S SCIENTISTS AND ONE OF
CHEMETICS, 75% OF THE ELECTROLYTE
RESISTANCE WAS OVERCOME WITH
MHD**

**IN ANOTHER SERIES OF HYDROGEN
EVOLUTION EXPERIMENTS AT FLEXIBLE
SOLUTIONS' LABORATORIES AS ABOVE,
SCIENTISTS OF DEFENCE RESEARCH
BOARD CANADA RECORDED AGAIN 75%
LESS ELECTROLYTE RESISTANCE IN
THE MAGNETIZED CELL**

Energy saving magnetohydrodynamics

IN ELECTROWINNING, ELECTROREFINING OR ELECTROLYTIC PROCESSES GENERALLY, FEAR OF FRACTALS (OUT GROWTH OF E.G. “TREES”) HAS RESULTED IN LARGE SEPARATION OF ELECTRODES, HENCE HIGHER RESISTANCES DUE TO GREATER PATH LENGTH AND ENERGY WASTAGE ON LOW CONDUCTIVITY ELECTROLYTE COMPARED TO METAL AND SEMI-CONDUCTORS IN THE CELL.

APPLICATIONS:

UTILITY LOAD SMOOTHING – THE 11 P.M.
DECLINE IN POWER DEMAND ... CHARGE
MAGNETIZED BATTERIES AGAINST 5 P.M.
SURGE, OR ELECTROLYZE WATER FOR USE
IN FUEL CELLS AT 5 P.M.

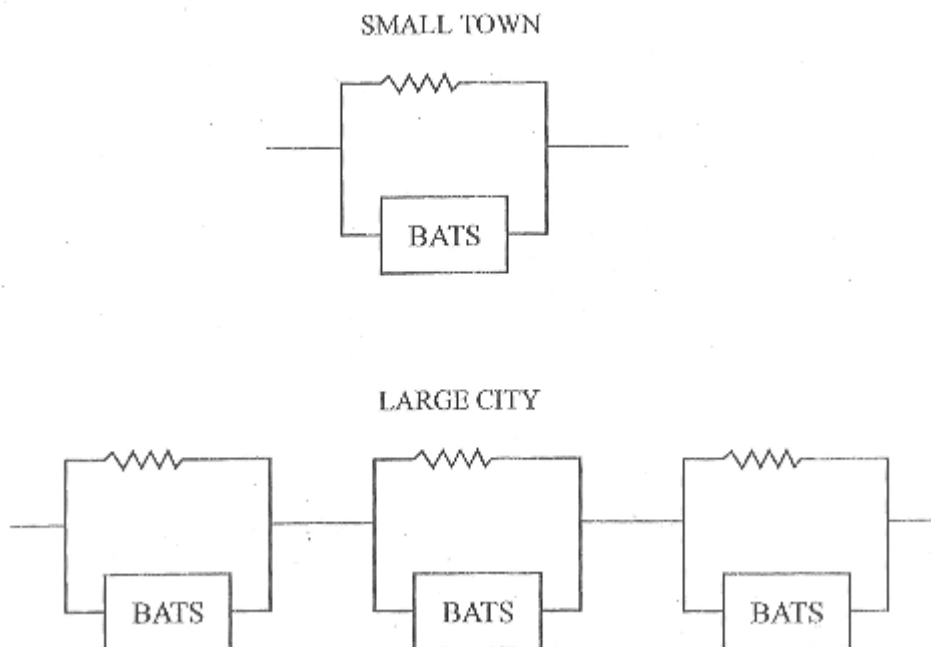
INTERMITTENT GENERATION: WIND AND SOLAR
GENERATION, CHARGE BATTERIES OR
GENERATE HYDROGEN AND OXYGEN AS
STORAGE AT PEAK PRODUCTION TO DRAW
ON IN THE DARK AND CALM DAYS.

TRACTION BATTERIES FOR HYBRID VEHICLES,
ROADABLE OR NOT: SINCE THE CHARGING
OF THE BATTERIES WILL BE BY GASOLINE
MOTOR, IF THE SUPPLIED ENERGY IS
STORED AT 90% AND THE STORED ENERGY
CAN BE SUPPLIED TO THE WHEELS AT 90%,
WHAT WOULD A CAR GETTING 50 MILES TO
THE GALLON EXPERIENCE IN INCREASED
MILEAGE?

...75 MILES TO THE GALLON? ... MORE?

GENERAL UTILITY LOAD SMOOTHING

WITH ONLY 3 TIME ZONES ACROSS THE USA, THE GENERAL HABITS OF THE ELECTRIC UTILITY USERS BECOMES IMPORTANT. WHEN EVERYONE GOES TO BED AND MOST ELECTRICAL EQUIPMENT IS SHUT OFF – EVEN THE AIR CONDITIONER USES LESS POWER AS THE NIGHT COOLS. THE GRID HAS MORE POWER AVAILABLE THAN CAN BE SHUNTED ON INTO THE NEXT TWO TIME ZONES. THERE ARE NOT THREE, BUT MORE THAN SIX HOURS, SO A STORAGE DEVICE FROM WHICH POWER CAN BE TAKEN FOR THE PEAK PERIODS OF FIVE PM TO EIGHT PM IS SHOWN, PLUS THE LOWERING OF LINE RESISTANCE LOSSES POSSIBLE WITH A BANK OF BATTERIES.



MAGNETIC FIELD ALTERATION IN MAGNETIZING A FOUR ELECTRODE BATTERY

Fig. 1

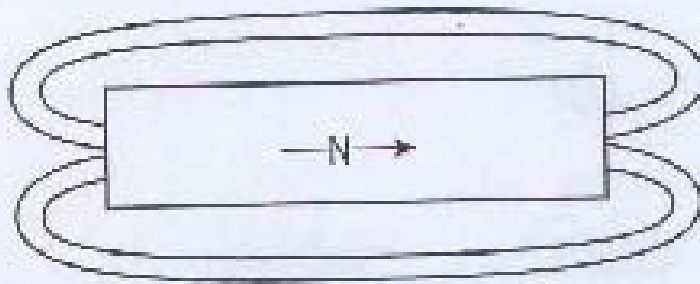


Fig. 2

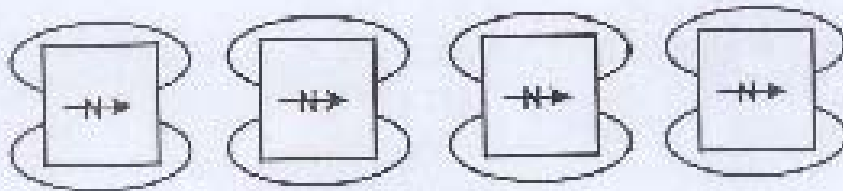


Fig.3

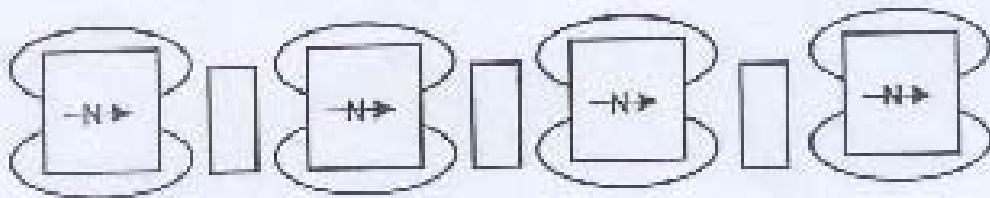
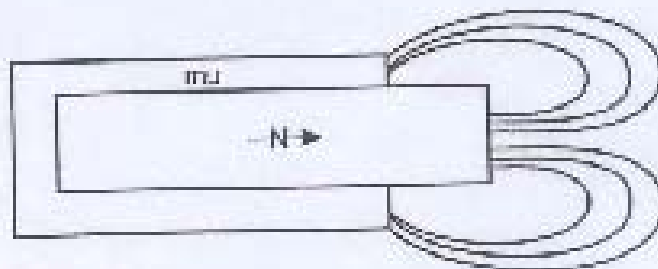


Fig. 4



MAGNETIC FIELD ALTERATION IN MAGNETIZING A FOUR ELECTRODE BATTERY

If a bar magnet (Fig. 1) is cut into four pieces (Fig. 2), four smaller, lower Magnetic field magnets are produced.

This is analogous to magnetizing a four Electrode battery, except that they are separated by electrolyte (Fig. 3) which further constricts the projected field and its strength.

Fig. 4 shows how by wrapping the bar magnet of Figure 1 in mu metal or soft iron, the field can be extended /projected further into the surrounding medium.

Fig.4 applies directly to the O'Brien magnetizer but is not used in the batteries studied.

MAGNETOHYDRODYNAMICS HAS BEEN TESTED IN NEW WORLD BATTERIES LABORATORIES FOR ELECTROWINNING, ELECTROREFINING, ELECTROPOLISHING AND ELECTROMACHINING. IT IS PLANNED TO DEMONSTRATE THESE TO SUITABLE AUDIENCES IN THE NEAR AND MIDTERM FUTURE.

UNTESTED FURTHER LIKELY CANDIDATES FOR BETTER PERFORMRANCE WITH MAGNETOHYDRODYNAMICS: ANY ELECTROCHEMICAL PROCESS FOR PRODUCING PRODUCTS OTHER THAN SODIUM CHLORATE. PRECURSORS FOR SOME PLASTICS ARE PRODUCED IN TONNAGES SOME PHARMACEUTICALS PRODUCED BY FERMENTATION PRODUCE RACEMATES WHICH ARE RESOLVED ELECTROCHEMICALLY. WE HAVE PRODUCED ADHERANT CONDUCTIVE COATINGS BY ELECTROCHEMICAL POLYMERIZATION.

THE LIST IS LONGER THAN THIS.

PATENT PROTECTION

1. **US5,051,157 (09/24/91)** Aimed at the lead acid battery, but also electrowinning, electroplating, electropolishing and electromachining, the technology since much improved.
2. **US6,194,093 B1 (03/25/01)** Four of the most common batteries, lead-acid, NiCd, NiMH and Zn-air covered for magnetohydrodynamics (MDH) improvement in charging and efficiency in discharging.
3. **US6,556,424 (04/29/03)** Magnetized supercapacitor aimed at passing power in hybrid autos or fuel cell cars.
4. **US6,741,440 B2 (05/25/04)** The O'Brien magnetizerTM, the method of magnetizing electrodes for MHD in batteries, reciprocating; now superseded by a new application for a patent for a rotating version for larger electrodes and more rapid magnetization.
5. **US7,045, 042, H₂ by water electrolysis, (05/16/05).** A method of eliminating about 75% of the resistance of the electrolyte. Aimed at fuel cell cars.
6. **US6,836,098 (09/30/04)** Uses both #'s 2 and 3 to effect a fast charge with no heating or deterioration of the batteries or supercapacitor. Accommodates regenerative braking in fuel cell cars and hybrids safely.

Regular patent applications are in for a number of advances, including one for Li-ion in which the generation of heat is reduced to almost the level of water electrodes and charging is faster and more energy efficient.

Another on a switching device that can be switched to fast charging for control of regenerative braking, and to recharging of the supercapacitor and magnetized batteries while being driven.

Other applications: Magnetized reverse osmosis application prevents fouling of the membranes, magnetized rapid electrochemical resolving of a racemate in pharmaceuticals, more rapid production of electrochemical products such as in precursors.

SPECIFIC ENERGY VS SPECIFIC POWER

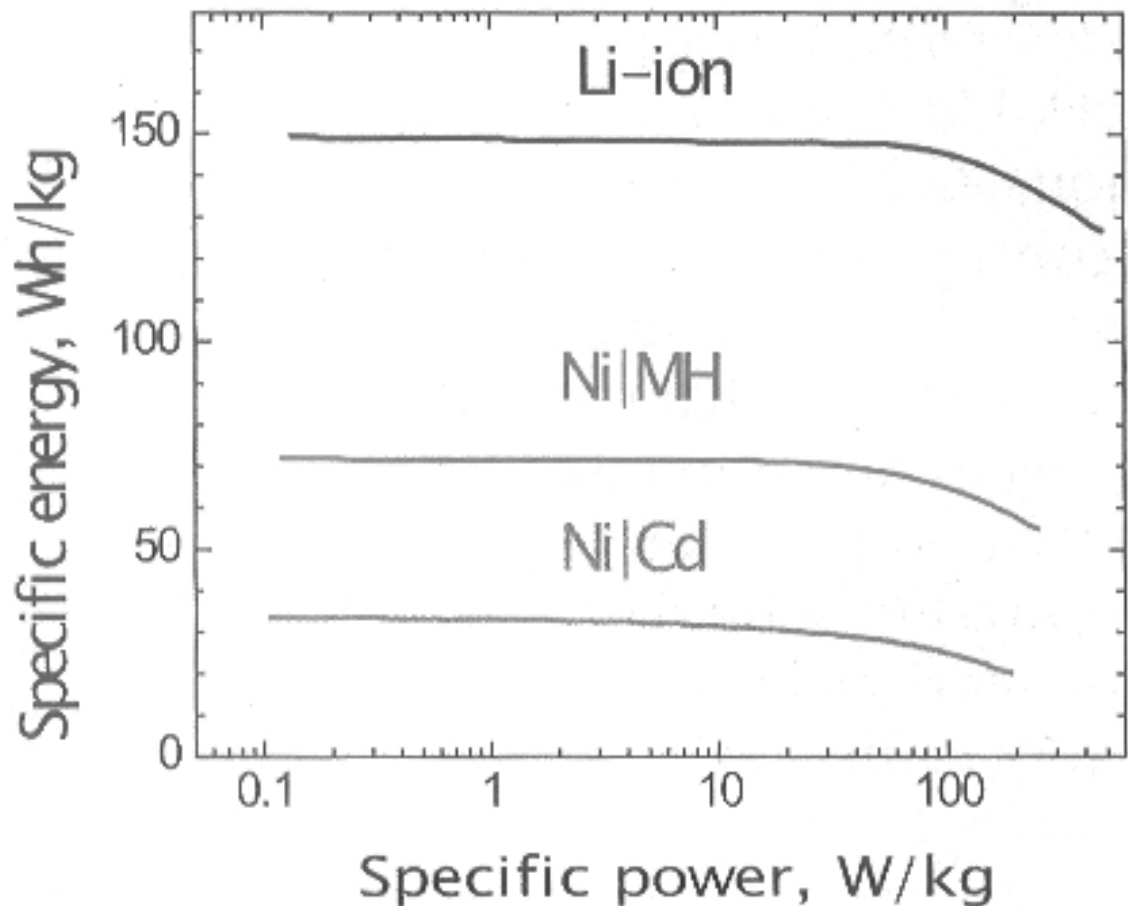
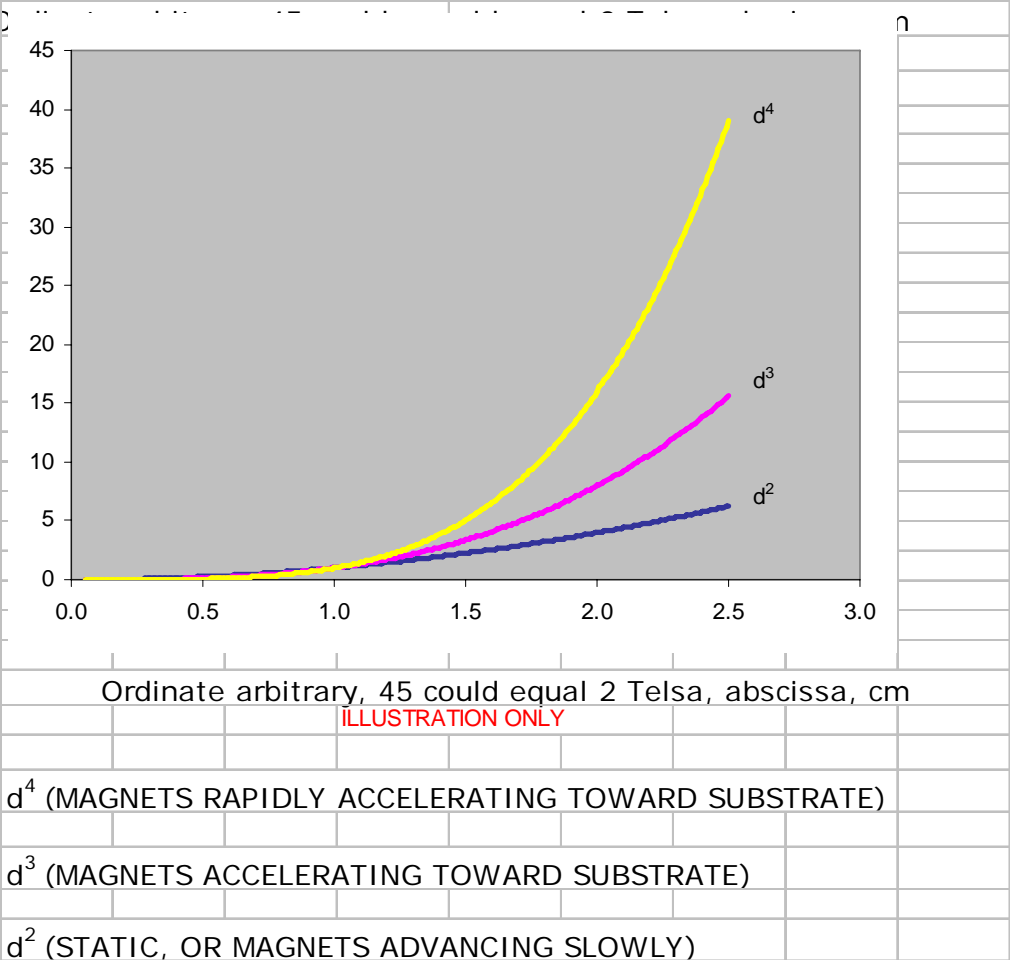
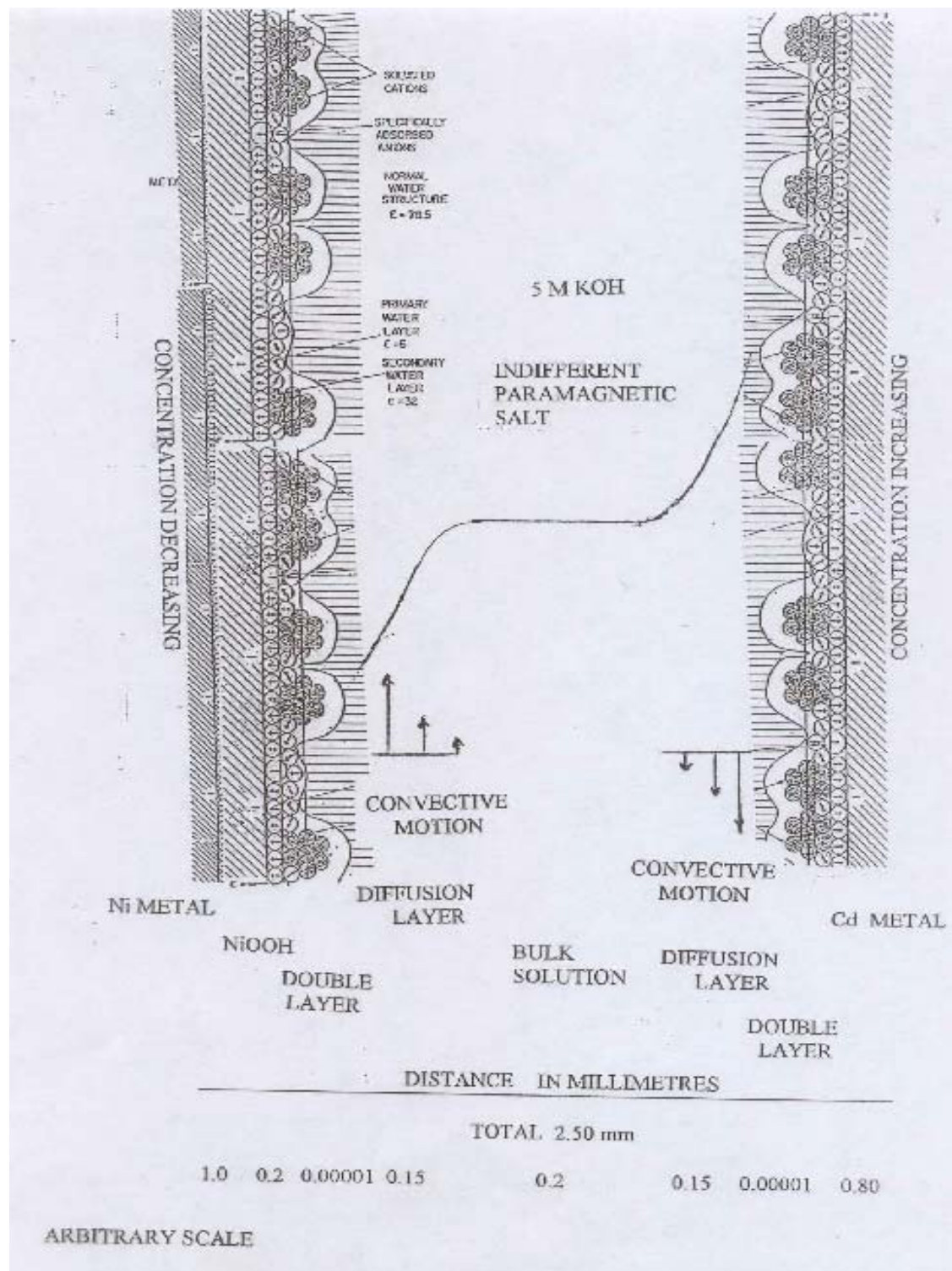


FIG. 3. Specific energy vs. specific power or Ragone plots, for three common rechargeable batteries. Ref: Handbook of batteries, D. Linden, T. B. Reddy, Eds., McGraw-Hill, New York (2002), 3rd Edition.

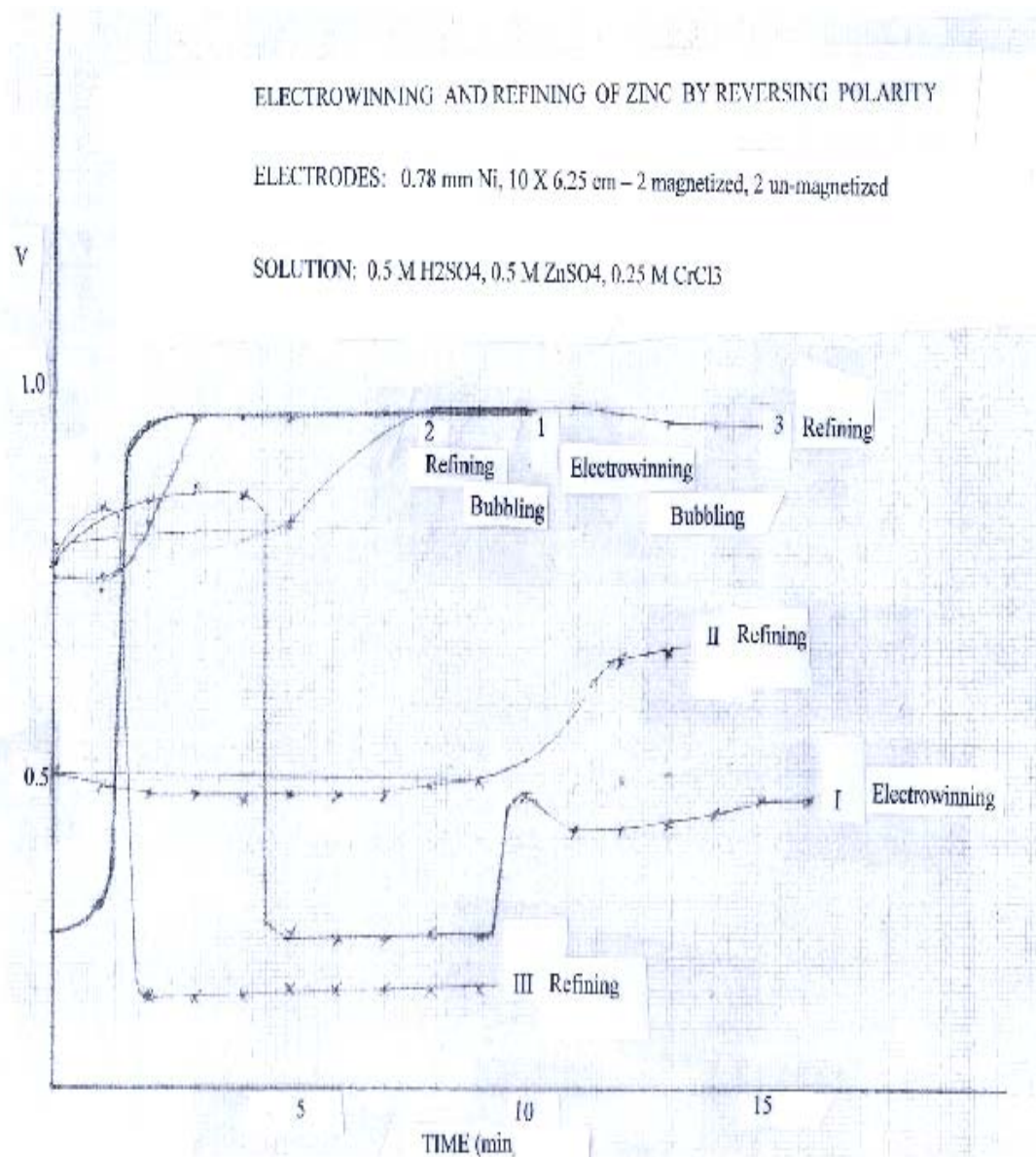
MAGNETIC HAMMERING



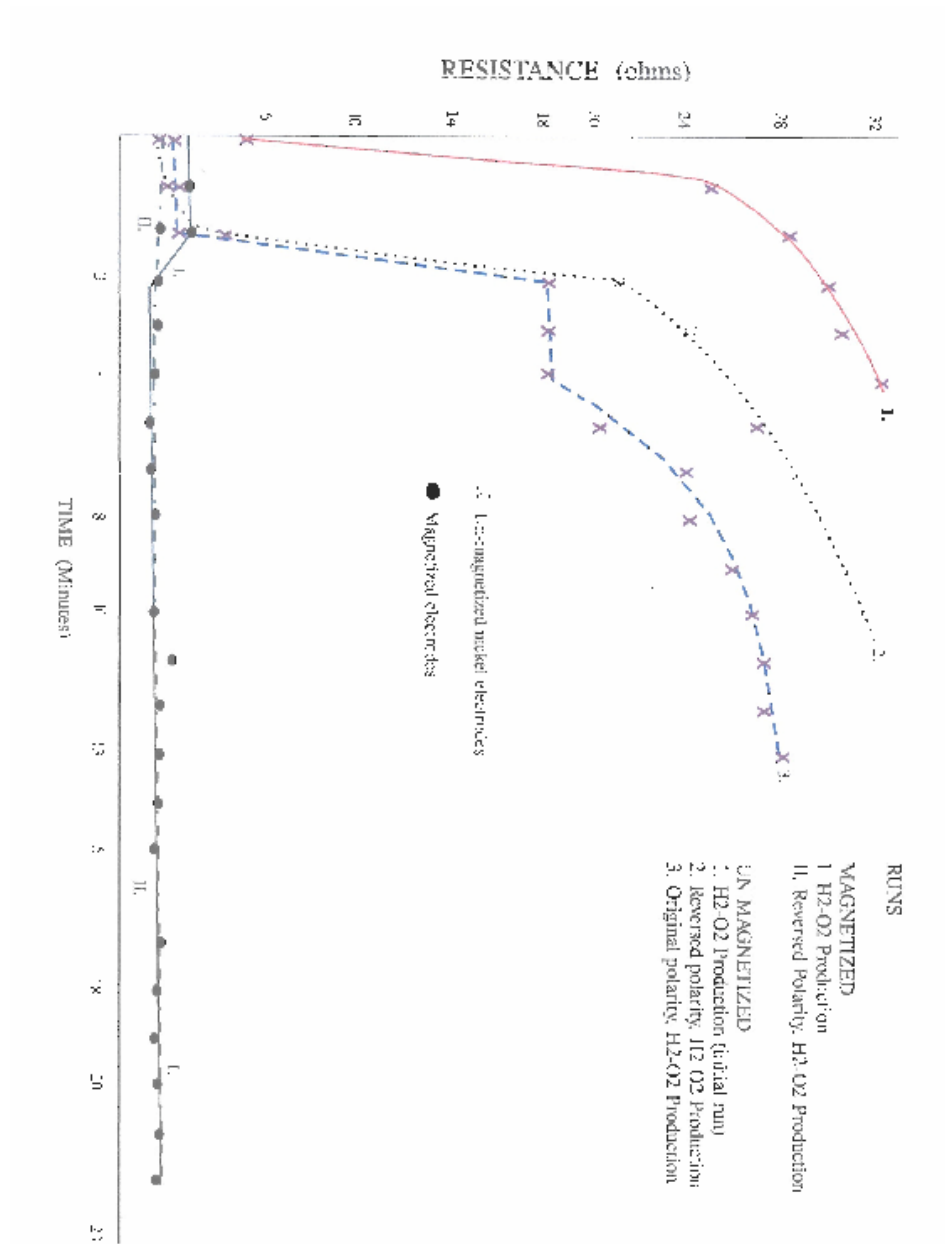
ARBITRARY SCALE



Electrowinning and Refining of Zinc



Resistance vs. time



Hydrogen Generated Electrolysis of Water

